

Code: 23BS1203

I B.Tech - II Semester – Regular Examinations - JULY 2024**ENGINEERING PHYSICS**
(Common for EEE, ECE, CSE)

Duration: 3 hours

Max. Marks: 70

- Note: 1. This question paper contains two Parts A and B.
 2. Part-A contains 10 short answer questions. Each Question carries 2 Marks.
 3. Part-B contains 5 essay questions with an internal choice from each unit. Each Question carries 10 marks.
 4. All parts of Question paper must be answered in one place.

BL – Blooms Level

CO – Course Outcome

PART – A

| | | BL | CO |
|------|---|----|-----|
| 1.a) | Mention the characteristics of a laser. | L1 | CO2 |
| b) | Define Total Internal Reflection. | L1 | CO1 |
| c) | Define packing fraction. | L1 | CO1 |
| d) | Discuss any two applications of X-ray diffraction. | L2 | CO3 |
| e) | Define Di-electric polarization and Di-electric constant. | L1 | CO3 |
| f) | Write any two properties of anti-ferro magnetic materials. | L1 | CO3 |
| g) | An electron is bound in one-dimensional potential box of size 1×10^{-10} m. Find its energy value in the ground state. | L3 | CO5 |
| h) | Define Fermi energy. | L1 | CO5 |
| i) | Show the variation of Fermi level with temperature in n-type semiconductor. | L2 | CO4 |
| j) | What is an intrinsic semiconductor? | L1 | CO1 |

PART – B

| | | | BL | CO | Max. Marks |
|-----------------|----|---|----|-----|------------|
| UNIT-I | | | | | |
| 2 | a) | Demonstrate the construction and working of Ruby laser. | L3 | CO2 | 6 M |
| | b) | Explain pumping and illustrate various pumping mechanisms. | L4 | CO4 | 4 M |
| OR | | | | | |
| 3 | a) | Explain acceptance angle and numerical aperture. Derive the expression for acceptance angle. | L3 | CO2 | 7 M |
| | b) | Select and discuss any three general applications of optical fiber. | L4 | CO4 | 3 M |
| UNIT-II | | | | | |
| 4 | a) | Sketch the seven types of crystal systems with the Bravais Lattices. | L3 | CO3 | 7 M |
| | b) | Copper has FCC structure and the atomic radius is 0.1278 nm. Calculate the inter planar spacing for (2 1 2) planes. | L3 | CO3 | 3 M |
| OR | | | | | |
| 5 | a) | Explain Bragg's law of X-ray diffraction. | L3 | CO3 | 3 M |
| | b) | Explain powder method to determine the crystal structure. | L4 | CO5 | 7 M |
| UNIT-III | | | | | |
| 6 | | Explain electronic polarization. Calculate the electronic polarizability in a Di-electric material. | L3 | CO3 | 10 M |

| OR | | | | | |
|----------------|----|---|----|-----|------|
| 7 | a) | Explain the classification of magnetic materials into Dia, Para and Ferro magnetic materials. | L3 | CO3 | 6 M |
| | b) | Differentiate soft and hard magnetic materials. | L4 | CO5 | 4 M |
| UNIT-IV | | | | | |
| 8 | a) | Interpret Schrodinger time independent wave equation for a particle. | L3 | CO3 | 7 M |
| | b) | Explain the significance of wave function. | L4 | CO5 | 3 M |
| OR | | | | | |
| 9 | a) | Explain Fermi-Dirac distribution function and its variation with temperature. | L4 | CO5 | 7 M |
| | b) | Explain the salient features of classical free electron theory of metals. | L3 | CO3 | 3 M |
| UNIT-V | | | | | |
| 10 | | Calculate the density of electrons in the conduction band of an intrinsic semiconductor. | L3 | CO2 | 10 M |
| OR | | | | | |
| 11 | a) | Explain Hall effect and derive the expression for Hall coefficient. | L4 | CO4 | 7 M |
| | b) | Infer any three applications of Hall effect. | L4 | CO4 | 3 M |